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A Concise Total Synthesis of Melithiazole C[†]

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ABSTRACT

A short and convergent synthesis of the myxobacterial antibiotic melithiazole C is described featuring a highly E-selective cross-metathesis as the key step.

The thiazole subunit is found in many natural and synthetic compounds with a wide range of interesting bioactivities.¹ The melithiazoles A–N represent a group of fungicidal β -methoxy acrylate (MOA) metabolites which have been isolated from different strains of myxobacteria.² As a common structural motif they share either a bisthiazole or a thiazole—thiazoline ring system linked via a double bond to a highly pharmacophoric polypropionate moiety that selectively inhibits mitochondrial respiration by binding to the cytochrome bc_1 complex (Figure 1).³

Figure 1. Melithiazoles A-N.

Isolated in only minor quantities from *Melittangium lichenicola*, melithiazole C (1) contains a single thiazole ring

and additionally occurs as a mixture of E/Z isomers at C6—C7 $(6E/6Z = 7/2)^2$ with only the *trans*-isomer being biologically active.⁴ On the other hand, derivatization of the acetyl group in **1** was found to selectively enhance its biological activities.⁴ Whereas simple transformation of the carbonyl group into an oxime functionality reduces the antifungal properties and increases the cytotoxicity, an exceptionally high antifungal activity is observed with a methoxime or vinyl derivative. Thus, melithiazole C (**1**) has attracted great attention as a potential lead for the development of agrochemical fungicides.⁵

Despite its very limited natural availability arising from the low efficiency of the fermentation process, only one total synthesis of **1** has been reported proceeding via a moderately stereoselective Wittig reaction ($E/Z \approx 6/1$) to introduce the C6–C7 double bond (14 steps, 15% overall yield).⁶ On the basis of our recent work on the cross-metathesis (CM) of vinyl substituted thiazoles,⁷ we present herein a short and

[†] Dedicated to the memory of Charles Mioskowski (1946-2007).

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highly stereoselective synthesis of **1** that, at the same time, should allow rapid access to structural derivatives^{5c} as well as various heterocyclic analogues.^{5b}

Retrosynthetically, **1** was therefore fragmented into the common β -methoxy acrylate **2** and the vinyl thiazole **3** which would both be readily accessible in a few steps from the commercially available Evans' propionate **4** and the dibromothiazole **5**, respectively (Scheme 1).

As depicted in Scheme 2, the synthesis of the β -methoxy acrylate **2** began with a previously described asymmetric aldol reaction between **4** and acrolein (dr >95:5)⁸ followed by *O*-methylation with methyl triflate in the presence of 2,6-ditertbutylpyridine (DTBP). While initial attempts to directly

transform the resulting oxazolidinone **7** into the required β -keto ester **8** under Reformatsky conditions failed (BrCH₂-CO₂Me, Zn, THF, reflux), activation of the corresponding carboxylic acid with use of carbonyl diimidazole (CDI) and direct condensation with the lithium enolate of methyl acetate at -78 °C in THF proceeded smoothly to produce **8** in 75% yield (3 steps). Finally, acid-catalyzed methyl enol ether formation with MeOH in the presence of trimethyl ortho-

formate gave the desired β -methoxy acrylate **2** as a single stereoisomer in 50% overall yield from **4**. ¹⁰

Concerning the preparation of the vinyl thiazole **3**, the thiazolyl bromide **9** was considered a suitable precursor since it is readily available from 2,4-dibromothiazole (**5**) via regioselective metalation (*n*-BuLi, Et₂O, -78 °C) and treatment with *N*-acetylmorpholine.¹¹ In our approach, a change of solvent from Et₂O to THF furnished the known bromide **9** in a significantly improved yield (81% vs 66%).¹¹ The latter was then subjected to a Stille cross-coupling reaction with vinyl tributyltin (1.1 equiv) under standard conditions [2 mol % of PdCl₂(PPh₃)₂, dioxane, 100 °C]¹² to efficiently afford the desired vinyl thiazole **3** (Scheme 3).

With both fragments in hand, the conditions for the final metathesis reaction were investigated. While Grubbs first generation catalyst (G I)¹³ proved completely ineffective (Table 1, entry 1), CM of an equimolar mixture of 2 and 3

Table 1. Optimization of the CM Reaction

entry	[Ru]	mol %	equiv of 3	$\mathrm{solvent}^a$	temp (°C)	time (h)	$\begin{array}{c} {\rm conversion}^{b,c} \\ (\%) \end{array}$
1	GΙ	10	1	$\mathrm{CH_{2}Cl_{2}}$	40	48	0
2	GII	10	1	$\mathrm{CH_2Cl_2}$	40	48	27
3	H-G	10	1	$\mathrm{CH_2Cl_2}$	40	48	18
4	$_{ m G~II}$	20	2	$\mathrm{CH_2Cl_2}$	40	60	50
5	$_{ m G~II}$	20	2	C_6H_6	60	60	26
6	$_{ m G~II}$	20	2	$\mathrm{CH_2Cl_2}$	20	72	5
7	$_{ m G~II}$	30	2	$\mathrm{CH_2Cl_2}$	40	60	$68 (56)^d$
8	$_{ m GII}$	40	2	CH_2Cl_2	40	60	72

 a 0.05 M. b Ratio of 1:2 determined by $^1{\rm H}$ NMR. c $E/Z \ge 20/1$. d Isolated yield.

in refluxing CH₂Cl₂ with 10 mol % of Grubbs second generation catalyst (G II)¹⁴ or with Hoveyda—Grubbs catalyst (H-G)¹⁵ resulted mainly in homodimerization of the vinyl

3426 Org. Lett., Vol. 9, No. 17, 2007

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thiazole 3 while 2 was consumed only slowly (Table 1, entries 2 and 3).

Since protected allylic alcohols such as **2** do not undergo any self-metathesis (Type III olefins),¹⁶ better results were obtained under more forcing conditions by employing 20 mol % of G II and an excess of thiazole **3**, whereas higher or lower temperatures were detrimental (Table 1, entries 4–6). Finally, CM was best effected with 30 mol % of G II in the presence of 2 equiv of **3** at 40 °C in CH₂Cl₂ (Table 1, entry 7) affording melithiazole C (**1**) as a single diastereo-isomer (E/Z > 20/1) in 56% isolated yield. The spectroscopic and physical data of **1** were identical with those reported for the natural product {[α]²⁰_D +167 (c 0.3, MeOH); lit. $[\alpha]^{22}_D$ +169 (c 0.3, MeOH)}.

In conclusion we have described a short and convergent synthesis of melithiazole C (1), which was obtained in 6 steps and 28% overall yield starting from the commercially available Evans' propionate 4. The key feature of this novel approach consists of a highly *E*-selective CM reaction between a vinyl thiazole and a polypropionate fragment. Further applications of this strategy to the synthesis of related natural products are currently under investigation.

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Supporting Information Available: Experimental details and characterization data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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Org. Lett., Vol. 9, No. 17, 2007

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